### PATENT APPLICATION

### IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

In re application of

Masahide MOHRI, et al.

Continuation Application of

Appln. No.: 08/416,738

Group Art Unit: Not Assigned

Confirmation No.: Not Assigned

Examiner: Not Assigned

Filed: June 27, 2001

For: METAL OXIDE POWDER AND METHOD FOR THE PRODUCTION OF THE

SAME

### PRELIMINARY AMENDMENT

Commissioner for Patents Washington, D.C. 20231

Sir:

Prior to examination, please amend the above-identified application as follows:

### **IN THE SPECIFICATION:**

### Before the first line insert the following paragraph:

This is a continuation of Application No. 08/416,738 filed August 11, 1994, the disclosure of which is incorporated herein by reference. The international application to which benefit is claimed was not published under PCT Article 21(2) in English.

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Replace the first paragraph of page 13 with the following paragraph:

A suitable calcination temperature is not necessarily critical since it depends

on the kind of the intended metal oxide, the kinds and concentrations of the

hydrogen halide, the molecular halogen and the component prepared from the

molecular halogen and steam, or the calcination time. It is preferably from 500 to

1500°C, more preferably from 600 to 1400°C. When the calcination temperature is

lower than 500°C, a long time is necessary for calcination. When the calcination

temperature exceeds 1500°C, many agglomerated particles tend to be contained in

the produced metal oxide powder.

Replace TABLE 2 on Page 34 with the following:

# GGGGAGGG . CEE/GI

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Table 2

Oxide							Calci	natior	Calcination conditions		
		A	tmos	phere	gas (	Atmosphere gas (vol. %)			Gas intro-	Maintaining	
H	HCl	HBr	HF	$ m r$ $ m HF$ $ m Cl_2$	$ m N_2$	$N_2$ $H_2O$ $H_2$ Air	$ m H_2$		duction	temp. (°C)	time (min.)
									temp. (°C)		
-	100								Room temp.	800	30
	45					10		45	Room temp.	1100	30
	001								Room temp.	1100	30
H	100								800	1100	30
-	30				7 0				800	1100	30
-	30							20	800	800	30
				30	09	10			800	1100	30
+				100					800	1100	30
<del> </del>				30	09	10			800	1100	30
1								100	Room temp.	1100	180
<del> </del>								100	Room temp.	1100	180

### IN THE CLAIMS:

Cancel claims 2, 3, 29 and 30.

### Please enter the following amended claims:

1 (amended). A metal oxide powder except  $\alpha$ -alumina, comprising polyhedral particles having at least 6 planes each, a number average particle size of from 0.1 to 300  $\mu$ m, and a  $D_{90}/D_{10}$  ratio of 5 or less where  $D_{10}$  and  $D_{90}$  are particle sizes at 10% and 90% accumulation, respectively from the smallest particle size side in a cumulative particle size curve of the particles, and

wherein a ratio of agglomerated particle size to a primary particle size is from 1 to 6.

4 (amended). The metal oxide powder according to claim 3, wherein said ratio of an agglomerated particle size to a primary particle size is from 1 to 3.

5 (amended). The metal oxide powder according to any one of claims 1 or 4, wherein said metal oxide is a simple metal oxide of a metal element selected from the group consisting of the metal elements of the Groups Ib, II, III, IV, V, VI, VII and VIII of the Periodic Table, except α-alumina powder.

6 (amended). The metal oxide powder according to any one of claims 1 or 4,

wherein said metal oxide is a simple metal oxide titanium.

7 (amended). The metal oxide powder according to any one of claims 1 or 4, wherein said metal oxide is a simple metal oxide of a metal selected from the group consisting of magnesium, zirconium and iron.

8 (amended). The metal oxide powder according to any one of claims 1 or 4, wherein said metal oxide is a simple metal oxide of cerium.

9 (amended). The metal oxide powder according to any one of claims 1 or 4, wherein said metal oxide is a simple metal oxide of a metal selected from the group consisting of indium and tin.

10 (amended). The metal oxide powder according to any one of claims 1 or 4, wherein said metal oxide is a simple metal oxide of a metal selected from the group consisting of zinc, cadmium, gallium, germanium, niobium, tantalum, antimony, bismuth, chromium, molybdenum, manganese, cobalt, nickel and uranium.

11 (amended). A rutile type titanium oxide powder comprising polyhedral particles each having at least 8 planes, a number average particle size of from 0.1 to 300  $\mu$ m, a  $D_{90}/D_{10}$  ratio of 5 or less where  $D_{10}$  and  $D_{90}$  are particle sizes at 10% and

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90% accumulation, respectively from the smallest particle size side in a cumulative

particle size curve of the particles, and a ratio of agglomerated particle size to

primary particle size of the particles is from 1 to 6.

28 (amended). The method according to claim 13 or 14, wherein said metal

oxide powder or metal oxide precursor powder is a metal oxide powder or metal

oxide precursor powder of a metal selected from the group consisting of magnesium,

titanium, and iron.

### **REMARKS**

Entry and consideration of this Preliminary Amendment is respectfully requested.

Respectfully submitted,

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### **APPENDIX**

## VERSION WITH MARKINGS TO SHOW CHANGES MADE

### IN THE SPECIFICATION:

The specification is changed as follows:

Before the first line the following sentence is inserted:

This is a continuation of Application No. 08/416,738 filed August 11, 1994, the disclosure of which is incorporated herein by reference. The international application to which benefit is claimed was not published under PCT Article 21(2) in English.

# Page 13, first paragraph, is replaced with the following paragraph:

A suitable calcination temperature is not necessarily critical since it depends on the kind of the intended metal oxide, the kinds and concentrations of the hydrogen halide, the molecular halogen and the component prepared from the molecular halogen and steam, or the calcination time. It is preferably from 500 to 1500°C, more preferably from 600 to 1400°C. When the calcination [time] temperature is lower than 500°C, a long time is necessary for calcination. When the calcination temperature exceeds 1500°C, many agglomerated particles tend to be contained in the produced metal oxide powder.

## The Table 2 on Page 34 is amended as follows:

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Table 2

No. Ho Ho 11 Ti02 1 12 Ti02 4 14 Ti02 1 15 Ti02 1 15 Ti02 1	HCI HB <sub>1</sub> 100 100 100	Atmosphere r HF Cl2	ااا	738 (1	vo.]. %)		( = = = = = = = = = = = = = = = = = = =		INTERTRICALITIES
Ti02 Ti02 Ti02 Ti02	HB	HH	Cl2	7 276			Gas intro-	Manicanting	
Ti02 Ti02 Ti02 Ti02			1 1	$N_2$	$H_2O$ $H_2$	 Air	duction	temp. (°C)	time (min.)
Ti02 Ti02 Ti02 Ti02	00 100 100						temp. (°C)		
Ti02 Ti02 Ti02 Ti02	100						Room temp.	800	30
Ti02 Ti02 Ti02	001				10	45	Room temp.	1100	30
Ti0 <sub>2</sub>	00	-			2		Room temp.	1100	30
Ti02	2						800	1100	30
601.	200			7.0			800	1100	30
201	30					70	800	800	30
71102	30		06	60	10		800	1100	30
		+	90	3	) TO	[45]	800	1100	30
			20	09	10	[71]	800	1100	30
			2	3	21	100	Roo	1100	180
C. 1 1102						100		1100	180

### IN THE CLAIMS:

Claims 2, 3, 29 and 30 are canceled.

The claims are amended as follows:

1 (amended). A metal oxide powder except  $\alpha$ -alumina, comprising polyhedral particles having at least 6 planes each, a number average particle size of from 0.1 to 300  $\mu$ m, and a D<sub>90</sub>/D<sub>10</sub> ratio of [10]  $\underline{5}$  or less where D<sub>10</sub> and D<sub>90</sub> are particle sizes at 10% and 90% accumulation, respectively from the smallest particle size side in a cumulative particle size curve of the particles, and

wherein a ratio of agglomerated particle size to a primary particle size is from 1 to 6.

4 (amended). The metal oxide powder according to claim 3, wherein said ratio of [a primary] an agglomerated particle size to [an agglomerated] a primary particle size is from 1 to 3.

5 (amended). The metal oxide powder according to any one of claims 1 [to] <u>or</u> 4, wherein said metal oxide is a simple metal oxide of a metal element selected from the group consisting of the metal elements of the Groups Ib, II, III, IV, V, VI, VII and VIII of the Periodic Table, except α-alumina powder.

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6 (amended). The metal oxide powder according to any one of [clams] claims

1 [to] or 4, wherein said metal oxide is a simple metal oxide titanium.

7 (amended). The metal oxide powder according to any one of claims 1 [to] or 4,

wherein said metal oxide is a simple metal oxide of a metal selected from the group

consisting of magnesium, zirconium and iron.

8 (amended). The metal oxide powder according to any one of claims 1 [to] or 4,

wherein said metal oxide is a simple metal oxide of cerium.

9 (amended). The metal oxide powder according to any one of claims 1 [to] or 4,

wherein said metal oxide is a simple metal oxide of a metal selected from the group

consisting of indium and tin.

10 (amended). The metal oxide powder according to any one of claims 1 [to]

or 4, wherein said metal oxide is a simple metal oxide of a metal selected from the

group consisting of zinc, cadmium, gallium, germanium, niobium, tantalum,

antimony, bismuth, chromium, molybdenum, manganese, cobalt, nickel and

uranium.

11 (amended). A rutile type titanium oxide powder comprising polyhedral

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particles each having at least 8 planes, a number average particle size of from 0.1 to

300 μm, a D<sub>90</sub>/D<sub>10</sub> ratio of 5 or less where D<sub>10</sub> and D<sub>90</sub> are particle sizes at 10% and

90% accumulation, respectively from the smallest particle size side in a cumulative

particle size curve of the particles, and a ratio of agglomerated particle size to

primary particle size of the particles is from 1 to 6.

28 (amended). The method according to claim 13 or 14, wherein said metal

oxide powder or metal oxide precursor powder is a metal oxide powder or metal

oxide precursor powder of a metal selected from the group consisting of magnesium,

titanium, [zirconium] and iron.